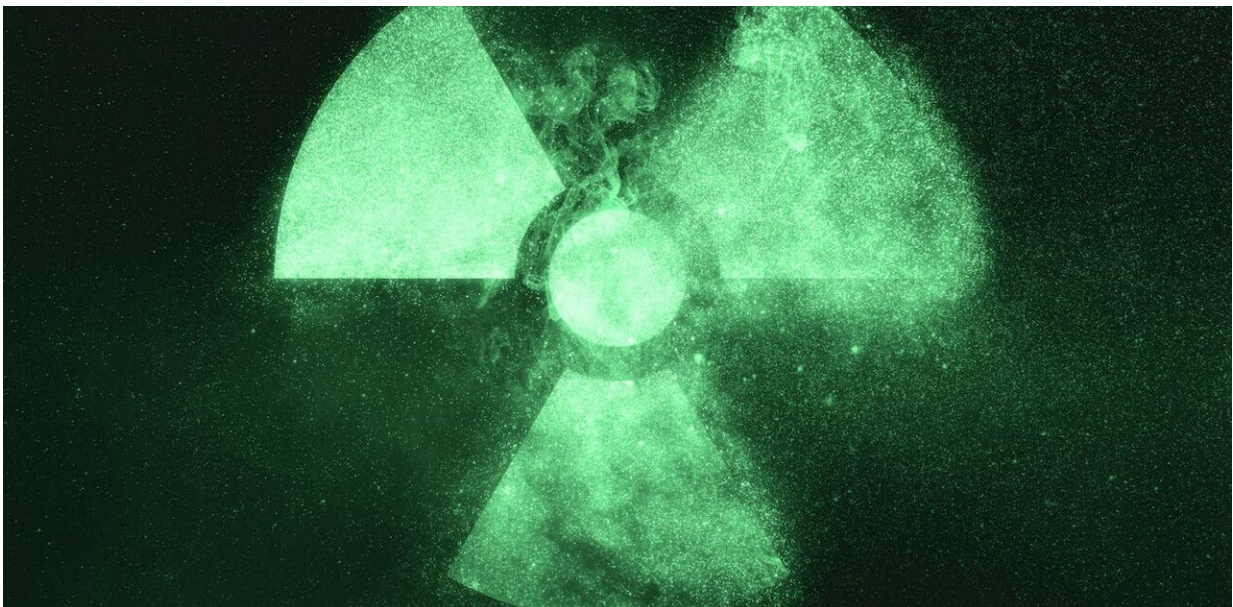


Are bananas really 'radioactive'? An expert clears up common misunderstandings about radiation

November 4 2022, by Sarah Loughran



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The simple mention of the word "radiation" often evokes fear in people. For others, it's fun to think a little exposure to radiation could turn you into the next superhero, just like [the Hulk](#).

But is it true basically everything around us is radioactive, even the food we eat? You may have heard bananas are mildly radioactive, but what

does that actually mean? And despite us not being superheroes, are [human bodies](#) also radioactive?

What is radiation?

[Radiation](#) is energy that travels from one point to another, either as waves or particles. We are exposed to [radiation](#) from various natural and artificial sources every day.

Cosmic radiation from the Sun and [outer space](#), radiation from rocks and soil, as well as radioactivity in the air we breathe and in our food and water, are all sources of natural radiation.

Bananas are a common example of a natural radiation source. They contain high levels of potassium, and a small amount of this is radioactive. But there's no need to give up your banana smoothie—the amount of radiation is extremely small, and far less than the natural "[background radiation](#)" we are exposed to every day.

Artificial sources of radiation include [medical treatments](#) and X-rays, mobile phones and power lines. There is a common misconception that artificial sources of radiation are more dangerous than naturally occurring radiation. However, this just isn't true.

There are no [physical properties](#) that make artificial radiation different or more damaging than natural radiation. The [harmful effects](#) are related to dose, and not where the exposure comes from.

What is the difference between radiation and radioactivity?

The words "radiation" and "radioactivity" are often used interchangeably. Although the two are related, they are not quite the

same thing.

Radioactivity refers to an unstable atom undergoing radioactive decay. Energy is released in the form of radiation as the atom tries to reach stability, or become non-radioactive.

The radioactivity of a material describes the rate at which it decays, and the process(es) by which it decays. So radioactivity can be thought of as the process by which elements and materials try to become stable, and radiation as the energy released as a result of this process.

Ionizing and non-ionizing radiation

Depending on the level of energy, radiation can be classified into two types.

[Ionizing radiation](#) has enough energy to remove an electron from an atom, which can change the chemical composition of a material. Examples of ionizing radiation include X-rays and radon (a radioactive gas found in rocks and soil).

[Non-ionizing radiation](#) has less energy but can still excite molecules and atoms, which causes them to vibrate faster. Common sources of non-ionizing radiation include mobile phones, power lines, and ultraviolet rays (UV) from the Sun.

Is all radiation dangerous? Not really

Radiation is not always dangerous—it depends on the type, the strength, and how long you are exposed to it.

As a general rule, the higher the energy level of the radiation, the more likely it is to cause harm. For example, we know that overexposure to

ionizing radiation—say, from naturally occurring radon gas—[can damage human tissues and DNA](#).

We also know that *non-ionizing* radiation, such as the UV rays from the Sun, can be harmful [if the person is exposed to sufficiently high intensity levels](#), causing [adverse health effects](#) such as burns, cancer, or blindness.

Importantly, because these dangers are well known and understood, they can be protected against. [International](#) and [national](#) expert bodies provide guidelines to ensure the safety and radiation protection of people and the environment.

For ionizing radiation, this means keeping doses above the natural background radiation as low as reasonably achievable—for example, only using medical imaging on the part of the body required, keeping the dose low, and retaining copies of images to avoid repeat exams.

For non-ionizing radiation, it means keeping exposure below [safety limits](#). For example, telecommunications equipment uses radiofrequency non-ionizing radiation and [must operate within these safety limits](#).

Additionally, in the case of UV radiation from the Sun, we know to [protect against exposure](#) using sunscreen and clothing when levels reach 3 and above on the UV index.

Radiation in medicine

While there are clear risks involved when it comes to radiation exposure, it's also important to recognize the benefits. One common example of this is the use of radiation in modern medicine.

[Medical imaging](#) uses ionizing radiation techniques, such as X-rays and

CT scans, as well as non-ionizing radiation techniques, such as ultrasound and magnetic resonance imaging (MRI).

These types of [medical imaging](#) techniques allow doctors to see what's happening inside the body and often lead to earlier and less invasive diagnoses. Medical imaging can also help to rule out serious illness.

Radiation can also help treat certain conditions—it can [kill cancerous tissue](#), shrink a tumor or even [be used to reduce pain](#).

So are our bodies also radioactive? The answer is yes, like everything around us, we are also a little bit radioactive. But this is not something we need to be worried about.

Our bodies were built to handle small amounts of radiation—that's why there is no danger from the amounts we are exposed to in our normal daily lives. Just don't expect this radiation to turn you into a superhero any time soon, because that definitely is science fiction.

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